

UNIVERSITY OF MADRAS
B.Sc. DEGREE PROGRAMME IN CHEMISTRY
 SYLLABUS WITH EFFECT FROM 2023-2024

Title of the Course	POLYMER SCIENCE						
Paper No.	EC VIII (EC-8)						
Category	Elective	Year	III	Credits	2	Course Code	324E6C
		Semester	VI				
Instructional hours per week	Lecture	Tutorial	Lab Practice		Total		
	4		-		4		
Prerequisites	Knowledge on functional groups and reaction mechanisms						
Objectives of the course	The course aims at providing an overall view of <ul style="list-style-type: none"> • classification of polymers, preparation of polymers • kinetics of polymerization and characterization of polymers • analytical techniques used to characterize polymers • reactions of polymers • speciality polymers like PVC, PMMA 						
Course Outline	UNIT I Introduction Difference between polymer and macromolecule – classification– synthetic and natural, organic and inorganic, thermoplastic and thermosetting. Plastics, elastomers, fibres and liquid resins.						
	Techniques of polymerization Bulk, solution, emulsion and suspension polymerization						
	Unit – II Kinetics of polymerization Kinetics of condensation and addition polymerisation; ionic, free radical, copolymerisation and coordination polymerisation – reactivity ratios – block and graft copolymers.						
	Characterisation of polymers Appearance, feel and hardness, density, effect of heat, solubility, combustion, tensile strength, shear, stress, impact strength, mechanical, thermomechanical and rheological properties of polymers in viscoelastic state.						

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	<p>UNIT III Molecular Weight and Properties of Polymers Molecular Weight of Polymers-Number Average and Weight Average, Molecular Weight Distribution, Determination of Molecular Weight polydispersity index – membrane and vapour phase osmometry, light scattering - Zimm plot, ultracentrifuge – sedimentation velocity and sedimentation equilibrium – viscometry – gel permeation chromatography Thermal properties of polymers – Glass Transition Temperature-State of Aggregation and State of Phase Transitions, Factors influencing Glass Transition Temperature, Importance of Glass Transition Temperature, Heat Distortion Temperature, TGA / DTA, Crystallinity of Polymers: Crystalline Behaviour, Degree of Crystallinity</p>
	<p>UNIT IV Reactions of Polymers-Hydrolysis, Acidolysis, Aminolysis, Addition and Substitution Reactions (One Example Each) Cyclisation, Cross-Linking and Reactions of Specific Functional Groups in the Polymer Polymer technology Processing of polymers – casting, thermoforming, moulding – extrusion, compression, blow moulding – foaming, lamination, reinforcing – processing of fibres – melt, wet and dry spinning.</p>
	<p>UNIT V Speciality polymers Polyelectrolytes, conducting polymers, polymeric supports for solid phase synthesis, biomedical polymers, liquid crystalline polymers, electroluminescent polymers – two examples of each of these polymers. Polyethylene, PVC, PMMA, polyester; rubber – synthetic and natural, vulcanisation of rubber. Polymer Degradation Types of Degradation - Thermal, Mechanical, Ultra Sound, Photo</p>
	<p>Radiation and Chemical Degradation Methods. Rubber-Natural and Synthetic-Structure, Mechanism of Vulcanisation Biodegradable and Non-Biodegradable Polymers.</p>
<p>Extended Professional Component (is a part of internal component only, Not to be included in the external examination question paper)</p>	<p>Questions related to the above topics, from various competitive examinations UPSC/ JAM /TNPSC others to be solved (To be discussed during the Tutorial hours)</p>

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Skills acquired from this course	Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills.
Recommended Text	<ol style="list-style-type: none"> Gowariker V.R, N.V. Viswanthan and Jayadev Sreedhar. Polymer Science. New Delhi: New Age International,2015 Misra G.S. Introductory Polymer Chemistry. New Delhi: Wiley Eastern,2010. Bahadur P and Sastry N V. Principles of Polymer Science. New Delhi: Narosa Publishing House,2005 Ahluwalia, V.K. Anuradha Mishra, <i>Polymer Science A Text Book</i>, Ane Books India: New Delhi,2008. Morrison, R. R.; Boyd, R. N.; Bhattacharjee, S. K. <i>Organic Chemistry</i>, 7thed.; Pearson: New Delhi,2011.
Reference Books	<ol style="list-style-type: none"> Billmeyer, F.W. Polymer Science. India: Wiley-Interscience,2007. Seymour, R. B.; Carraher Jr.C.E. Polymer Chemistry: An Introduction, Marcel Dckker Inc : New York, 1981. Sinha, R. Outlines of Polymer Technology, Prentice Hall of India: New Delhi,2000. Joel R. Fried, Polymer Science and Technology, 3rded.; Prentice Hall of India: New Delhi, 2014.
Website and e-learning source	<ol style="list-style-type: none"> https://polymerdatabase.com http://amrita.vlab.co.in/?sub=2&brch=190&sim=603&cnt=1 http://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/polymers.htm http://nsdl.niscair.res.in/bitstream/123456789/406/2/Molecular+weights+of+polymers.pdf
<p>Course Learning Outcomes (for Mapping with POs and PSOs) On completion of the course the students should be able to</p> <p>CO1: explain classification of polymers, elastomers, fibres and liquid resins</p> <p>CO2: explain addition and condensation polymerization, mechanical properties of polymers</p> <p>CO3: determine the molecular weight of polymers, and explain the thermal properties of polymers</p> <p>CO4: explain reactions of polymers and polymer processing</p> <p>CO5: discuss speciality polymers like PVC, PMMA, rubbers, biodegradable polymers</p>	

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10
CO1	S	S	S	S	S	S	S	M	S	M

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CO2	M	S	S	S	M	S	S	M	M	M
CO3	S	S	S	M	S	S	S	M	S	M
CO4	S	S	S	S	S	S	S	M	M	M
CO5	S	M	S	S	S	S	S	M	M	S

CO-PO Mapping (Course Articulation Matrix)

CO /PO	PO1	PO2	PO3	PO4	PO5
CO1	3	3	3	3	3
CO2	3	3	3	3	3
CO3	3	3	3	3	3
CO4	3	3	3	3	3
CO5	3	3	3	3	3
Weightage	15	15	15	15	15
Weighted percentage of Course Contribution to POs	3.0	3.0	3.0	3.0	3.0

Level of Correlation between PO's and CO's